Lovibond[®] Water Testing

Tintometer[®] Group



Safety Data Sheet

acc. to OSHA HCS (HazCom 2012)

Printing date 06/24/2022

1 Identification

· Product identifier

- Trade name: Peracetic acid Buffer HP1
- · Catalogue number: 56Z741598, 56L7415, 56L741530, 56L741565, 56U741530, 56U741565, SDT051
- · Application of the substance / the mixture: Reagent for water analysis
- · Manufacturer/Supplier: Tintometer Inc. 6456 Parkland Drive Sarasota, FL 34243 USA phone: (941) 756-6410 fax: (941) 727-9654 www.lovibond.us Made in Germany
- · Emergency telephone number: + 1 866 928 0789 (English, French, Spanish)

2 Hazard(s) identification

· Classification of the substance or mixture

GHS05 Corrosion

Corrosive to Metals 1 H290 May be corrosive to metals.

Skin Corrosion 1A H314 Causes severe skin burns and eye damage.

Eye Damage 1 H318 Causes serious eye damage.

· Label elements

- GHS label elements The product is classified and labeled according to the Hazard Communication Standard (HCS).
- · Hazard pictograms



- · Signal word Danger
- · Hazard-determining components of labeling:
- sulphuric acid 61 %
- Hazard statements
- H290 May be corrosive to metals.
- H314 Causes severe skin burns and eye damage.
- · Precautionary statements
- P260 Do not breathe mist/vapours/spray.
- Wear protective gloves/protective clothing/eye protection/face protection. P280
- P303+P361+P353 IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water or shower.
- P305+P351+P338 If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.
- P308 IF exposed or concerned: P310
 - Immediately call a poison center/doctor.

Other hazards

Contact with skin and inhalation of aerosols/ vapours of the preparation should be avoided. Acid burns have to treated immediately, as it may otherwise cause badly curing wounds.

Reviewed on 06/24/2022

Printing date 06/24/2022

Reviewed on 06/24/2022

Trade name: Peracetic acid Buffer HP1

(Contd. of page 1)

	_				
<u>_</u>		position/			
.5					
<u> </u>				In groc	

- Chemical characterization: Mixtures
- · Description: sulfuric acid solution
- Composition and Information on Ingredients:

Cancer Status IARC: Strong inorganic acid mists containing sulphuric acid can cause cancer.

Percent ranges are used due to the confidential product information.

CAS: 7664-93-9 EINECS: 231-639-5	sulphuric acid	♦ Corrosive to Metals 1, H290; Skin Corrosion 1A, H314	60–70%
Index number: 016-020-00-8 RTECS: WS5600000			
Additional informations For			

Additional information: For the wording of the listed hazard phrases refer to section 16.

4 First-aid measures

· Description of first aid measures

• General information: Immediately remove any clothing soiled by the product.

- · After inhalation:
- Supply fresh air.
- Call a doctor immediately.

• After skin contact:

Immediately wash with polyethylene glycol 400.

Immediately rinse with plenty of water.

Immediate medical treatment necessary. Failure to treat burns can prevent wounds from healing.

· After eye contact:

Rinse opened eye for several minutes (at least 15 min) under running water.

Call a doctor immediately.

· After swallowing:

Rinse out mouth and then drink 1-2 glasses of water.

Do not induce vomiting; immediately call for medical help.

- Most important symptoms and effects, both acute and delayed
- strong caustic effect after inhalation:
- coughing

breathing difficulty

- damage to the affected mucous membranes
- after swallowing:
- sickness
- vomiting
- diarrhoea

pain • **Danger:**

Danger of circulatory collapse. Danger of gastric perforation.

Danger of pulmonary edema.

Indication of any immediate medical attention and special treatment needed:

If swallowed or in case of vomiting, danger of entering the lungs. Later observation for pneumonia and pulmonary edema.

5 Fire-fighting measures

Extinguishing media

• Suitable extinguishing agents: CO₂, sand, extinguishing powder.

- For safety reasons unsuitable extinguishing agents:
- Water

--> exothermic reaction.

· Special hazards arising from the substance or mixture

The product is not combustible.

Formation of toxic gases is possible during heating or in case of fire.

In case of fire, the following can be released:

Sulfur oxides (SOx)

Printing date 06/24/2022

Reviewed on 06/24/2022

Trade name: Peracetic acid Buffer HP1

- · Advice for firefighters Protective equipment: Wear self-contained respiratory protective device. Wear fully protective suit. Additional information Collect contaminated fire fighting water separately. It must not enter the sewage system. Dispose of fire debris and contaminated fire fighting water in accordance with official regulations. Ambient fire may liberate hazardous vapours. 6 Accidental release measures Personal precautions, protective equipment and emergency procedures Advice for non-emergency personnel: Wear protective equipment. Keep unprotected persons away.
 - Avoid substance contact.
 - Ensure adequate ventilation
- Use respiratory protective device against the effects of fume/dust/aerosol. Advice for emergency responders: Protective equipment: see section 8
- · Environmental precautions: Do not allow product to reach sewage system or any water course.
- · Methods and material for containment and cleaning up:
- Ensure adequate ventilation.
- Use neutralizing agent.
- Absorb with liquid-binding material (sand, diatomite, universal binders).
- Dispose contaminated material as waste according to item 13.
- **Reference to other sections** See Section 8 for information on personal protection equipment. See Section 13 for disposal information.

7 Handling and storage

- · Precautions for safe handling
- · Advice on safe handling:
- Ensure good ventilation/exhaustion at the workplace.
- Prevent formation of aerosols.
- Hygiene measures:
- Do not inhale gases / fumes / aerosols. Do not get in eyes, on skin, or on clothing. Take off immediately all contaminated clothing.
- Wash hands before breaks and at the end of work.
- Do not eat, drink or smoke when using this product.
- · Conditions for safe storage, including any incompatibilities
- Requirements to be met by storerooms and receptacles: Store in a cool location.
- Keep only in original container.
- Information about storage in one common storage facility:
- Store away from metals. Do not store together with alkalis (caustic solutions).
- Store away from flammable substances. Further information about storage conditions:
- Keep receptacle tightly sealed.
- Protect from heat and direct sunlight.
- Protect from exposure to the light.
- Protect from humidity and water.
- · Recommended storage temperature: 20°C +/- 5°C (approx. 68°F)
- Specific end use(s) No further relevant information available.

(Contd. of page 2)

Printing date 06/24/2022

· Control parameters

Reviewed on 06/24/2022

(Contd. of page 3)

Trade name: Peracetic acid Buffer HP1

8 Exposure controls/personal protection

ſ	· Components with limit values that require monitoring at the workplace:			
ſ	CAS: 7664-93-9 sulphuric acid			
ſ	PEL (USA)	Long-term value: 1 mg/m³		
l	REL (USA)	Long-term value: 1 mg/m³		
		Long-term value: 0.2* mg/m³ *as thoracic fraction, A2		
		Long-term value: 0.2 mg/m³ thoracic, ACGIH A2; IARC 1		
l	EV (Canada)	Long-term value: 0.2 mg/m³		

· Additional information: The lists that were valid during the creation were used as basis.

Engineering measures:

Technical measures and appropriate working operations should be given priority over the use of personal protective equipment. See item 7.

· Personal protective equipment:

Protective clothing should be selected specifically for the workplace, depending on concentration and quantity of the hazardous substances handled.

- Breathing equipment: Use respiratory protective device against the effects of fume/dust/aerosol.
- · Recommended filter device for short term use: Combination filter B-P2
- · Protection of hands:
- Acid resistant gloves

Preventive skin protection by use of skin-protecting agents is recommended.

After use of gloves apply skin-cleaning agents and skin cosmetics.

- Material of gloves
- Fluorocarbon rubber (Viton) Butyl rubber, BR

Recommended thickness of the material: ≥ 0.35 mm

- Penetration time of glove material
- Value for the permeation: Level \leq 1 (10 min)

The exact break through time has to be found out by the manufacturer of the protective gloves and has to be observed.

- Eye protection:
- Tightly sealed goggles
- Use protective goggles that have been tested and approved in accordance with government standards (like NIOSH).
- · Body protection: Acid resistant protective clothing

· Limitation and supervision of exposure into the environment:

Do not allow product to reach sewage system or any water course.

nation on heale why deal and showled we wanted

9 Physical and chemical properties

· Information on basic physical and chemical properties				
· Appearance:				
 Form / Physical state: 	Solution			
· Color:	Yellow			
· Odor:	Odorless			
· Odor threshold:	Not applicable.			
· pH-value at 20°C (68°F):	<1			
	Strongly acidic			
 Melting point/freezing point: 	Not determined.			
Initial boiling point and boiling range:	Not determined.			
· Flash point:	Not applicable.			
• Flammability (solid, gas):	Not applicable.			
 Ignition temperature: 	Not applicable.			
 Decomposition temperature: 	Not determined.			
 Auto-ignition temperature: 	Product is not self-igniting.			
Danger of explosion:	Product does not present an explosion hazard.			
Flammability or explosive limits:				
Lower:	Not applicable.			

Printing date 06/24/2022

Reviewed on 06/24/2022

Trade name: Peracetic acid Buffer HP1

		(Contd. of page
· Upper:	Not applicable.	
Oxidizing properties:	CAS 7664-93-9 :	
	Oxidizing potential	
· Vapor Pressure:	Not determined.	
Density at 20°C (68°F):	1.56 g/cm³ (13.02 lbs/gal)	
· Relative density:	Not determined.	
· Vapor density:	Not determined.	
Evaporation rate:	Not determined.	
· Solubility(ies)		
· Water:	Fully miscible.	
· Partition coefficient (n-octand	ol/water): Not applicable (mixture).	
· Viscosity:		
· Kinematic:	Not determined.	
· Other information		
· Solids content:	< 5 %	
· Solvent content:		
· Organic solvents:	0 %	
· Water:	30-40 %	

10 Stability and reactivity

- · Reactivity see section "Possibility of hazardous reactions"
- · Chemical stability Stable at ambient temperature (room temperature).
- · Possibility of hazardous reactions

Corrosive action on metals. Reacts with metals forming hydrogen (Danger of explosion in case of large amounts!) When diluting, always add acid to water, never vice versa. Diluting or dissolving in water always causes rapid heating. Reacts with reducing agents. Reacts with peroxides. Reacts with halogenated compounds. Reacts with oxidizing agents. Reacts with acids and alkali (lyes). Reacts with ammonia (NH₃). · Conditions to avoid strong heating · Incompatible materials: metals combustible materials organic solvents Hazardous decomposition products: see section 5

11 Toxicological information

· Information on toxicological effects

· Acute toxicity: Based on available data, the classification criteria are not met.

· LD/LC50 values that are relevant for classification:			
CAS: 7664	CAS: 7664-93-9 sulphuric acid		
Oral	LD50	2140 mg/kg (rat) (IUCLID)	
Inhalative		510 mg/m³/2h (rat) IUCLID	

· Primary irritant effect:

· on the skin: Causes severe skin burns.

· on the eye:

Causes serious eye damage.

Risk of blindness!

· Sensitization: Based on available data, the classification criteria are not met.

(Contd. of page 5)

1

Κ

Safety Data Sheet acc. to OSHA HCS (HazCom 2012)

Printing date 06/24/2022

Reviewed on 06/24/2022

Trade name: Peracetic acid Buffer HP1

· Carcinogenic categories

· IARC (International Agency for Research on Cancer)

CAS: 7664-93-9 sulphuric acid

· NTP (National Toxicology Program)

CAS: 7664-93-9 sulphuric acid

· OSHA-Ca (Occupational Safety & Health Administration)

None of the ingredients is listed.

· Other information:

see section 8 / 15

Cancer Status of Sulfuric acid: The International Agency for Research on Cancer (IARC) has classified "strong inorganic acid mists containing sulfuric acid" as a known human carcinogen, (IARC category 1). This classification applies only to mists containing sulfuric acid and not to sulfuric acid or sulfuric acid solutions.

A2 (Suspected for humans) by ACGIH

· Synergistic Products: None

· CMR effects (carcinogenity, mutagenicity and toxicity for reproduction): The following statements refer to the mixture:

- · Germ cell mutagenicity Based on available data, the classification criteria are not met.
- Carcinogenicity Based on available data, the classification criteria are not met.
- · Reproductive toxicity Based on available data, the classification criteria are not met.
- STOT (specific target organ toxicity) -single exposure Based on available data, the classification criteria are not met. • STOT (specific target organ toxicity) -repeated exposure Based on available data, the classification criteria are not met.
- · Aspiration hazard Based on available data, the classification criteria are not met.

· Additional toxicological information:

In case of an acute molybdenum(VI) intoxication: diarrhoea, anaemia, fatigue, loss of appetite. Toxic effect on liver and kidneys after

high doses.

Swallowing will lead to a strong caustic effect on mouth and throat and to the danger of perforation of esophagus and stomach. The aerosol is corrosive to the eyes, the skin and the respiratory tract. Inhalation of aerosols may cause lung oedema.

CAS: 7664-93-9 sulphuric acid

(source: GESTIS)

Main toxic effects Acute: Irritation up to chemical burns to the mucous membranes and skin, danger of serious damage to the eyes and lungs Chronic: Irritation to the eyes and airways, erosion of the teeth, damage to the skin

Further Information:

Concentrated S. differs considerably from dilute Sulfuric acid with regard to chemical properties and effects. With increased dilution Sulfuric acid acts less aggressively.

12 Ecological information

· Toxicity · Aquatic toxicity: CAS: 7664-93-9 sulphuric acid EC50 >100 mg/l/48h (Daphnia magna) (OECD 202) (ECHA) LC50 16-29 mg/l/96h (bluegill) (Merck) Bacterial toxicity: sulfates toxic > 2.5 g/l Other information: Toxic for fish: sulfates > 7 g/l molybdenum compounds in general: > 25 mg/l $NH_{4}^{+} > 0.3 mg/l$ · Persistence and degradability . • Other information: Mixture of inorganic compounds.

Methods for the determination of biodegradability are not applicable to inorganic substances.

· Bioaccumulative potential No further relevant information available.

Printing date 06/24/2022

Reviewed on 06/24/2022

Trade name: Peracetic acid Buffer HP1

Mobility in soil No further relevant information available.
 Other adverse effects
 Harmful effect due to pH shift.

 Forms corrosive mixtures with water even if diluted.

Avoid transfer into the environment.

13 Disposal considerations

· Waste treatment methods

· Recommendation:

Must not be disposed of together with household garbage. Do not allow product to reach sewage system. Hand over to hazardous waste disposers.

· Uncleaned packagings:

- Recommendation: Disposal must be made according to official regulations.
- · Recommended cleansing agent: Water, if necessary with cleansing agents.

Transport information	
UN-Number DOT, IMDG, IATA	UN1830
UN proper shipping name DOT IMDG, IATA	Sulfuric acid mixture SULPHURIC ACID mixture
Transport hazard class(es)	
DOT CORROSIVE	
Class	8 Corrosive substances
Label	8
Class	8 Corrosive substances
Label	8
Packing group DOT, IMDG, IATA	II
Environmental hazards:	Not applicable.
Special precautions for user Hazard identification number (Kemler code): EMS Number: Segregation groups Stowage Category Stowage Code	Warning: Corrosive substances 80 F-A,S-B Acids E SW15 For metal drums, stowage category B.
Transport in bulk according to Annex II of MARPO and the IBC Code	L73/78 Not applicable.
Transport/Additional information:	
DOT Quantity limitations	On passenger aircraft/rail: 1 L On cargo aircraft only: 30 L

(Contd. of page 6)

(Contd. of page 7)

Safety Data Sheet acc. to OSHA HCS (HazCom 2012)

Printing date 06/24/2022

Reviewed on 06/24/2022

Trade name: Peracetic acid Buffer HP1

·IMDG

- · Limited quantities (LQ)
- Excepted quantities (EQ)

1L Code: E2 Maximum net quantity per inner packaging: 30 ml Maximum net quantity per outer packaging: 500 ml

15 Regulatory information

Safety, health and environmental regulations/legislation specific for the substance or mixture
 Sara
 Section 355 (Extremely hazardous substances):

CAS: 7664-93-9 sulphuric acid

• Section 313 (Specific toxic chemical listings):

CAS: 7664-93-9 sulphuric acid

• **TSCA (Toxic Substances Control Act):** All components have the value ACTIVE.

· Hazardous Air Pollutants

None of the ingredients is listed.

· Proposition 65

· Proposition 65		
· Chemicals known to cause cancer:		
None of the ingredients is listed.		
· Chemicals known to cause reproductive toxicity for females:		
None of the ingredients is listed.		
· Chemicals known to cause reproductive toxicity for males:		
None of the ingredients is listed.		
· Chemicals known to cause developmental toxicity:		
None of the ingredients is listed.		
· New Jersey Right-to-Know List:		
CAS: 7664-93-9 sulphuric acid		
· New Jersey Special Hazardous Substance List:		
CAS: 7664-93-9 sulphuric acid	CA, CO, R2	
· Pennsylvania Right-to-Know List:		
CAS: 7664-93-9 sulphuric acid		
· Pennsylvania Special Hazardous Substance List:		
CAS: 7664-93-9 sulphuric acid	E	
· EPA (Environmental Protection Agency)		
None of the ingredients is listed.		
 NIOSH-Ca (National Institute for Occupational Safety and Health) 		
None of the ingredients is listed.		
Information about limitation of use:		

Information about limitation of use:
 Observe national regulations where applicable:
 Employment restrictions concerning young persons must be observed.

· Chemical safety assessment: A Chemical Safety Assessment has not been carried out.

16 Other information

This information is based on our present knowledge. However, this shall not constitute a guarantee for any specific product features and shall not establish a legally valid contractual relationship.

Relevant phrases

H290 May be corrosive to metals.

H314 Causes severe skin burns and eye damage.

Printing date 06/24/2022

Reviewed on 06/24/2022

Trade name: Peracetic acid Buffer HP1

 Date of preparation / last revision 06/24/2022 / -Abbreviations and acronyms: STOT: specific target organ toxicity SE: single exposure RE: repeated exposure EC50: half maximal effective concentration IC50: hallf maximal inhibitory concentration NOEL or NOEC: No Observed Effect Level or Concentration ACGIH[®] - American Conference of Governmental Industrial Hygienists •A1 - Confirmed human carcinogen •A2 - Suspected human carcinogen •A3 - Confirmed animal carcinogen with unknown relevance to humans A4 - Not classifiable as a human carcinogen
 A5 - Not suspected as a human carcinogen
 IARC - International Agency for Research on Cancer
 Group 1 - Carcinogenic to humans •Group 2A - Probably carcinogenic to humans •Group 2B - Possibly carcinogenic to humans •Group 3 - Not classifiable as to carcinogenicity to humans •Group 4 - Probably not carcinogenic to humans NTP - National Toxicology Program, U.S. Department of Health and Human Services •Group K - Known to be Human Carcinogens •Group R - Reasonably Anticipated to be Human Carcinogens IMDG: International Maritime Code for Dangerous Goods DOT: US Department of Transportation IATA: International Air Transport Association EINECS: European Inventory of Existing Commercial Chemical Substances ELINCS: European List of Notified Chemical Substances CAS: Chemical Abstracts Service (division of the American Chemical Society) LC50: Lethal concentration, 50 percent LD50: Lethal dose, 50 percent NIOSH: National Institute for Occupational Safety OSHA: Occupational Safety & Health TLV: Threshold Limit Value PEL: Permissible Exposure Limit REL: Recommended Exposure Limit Corrosive to Metals 1: Corrosive to metals - Category 1 Skin Corrosion 1A: Skin corrosion/irritation - Category 1A Eye Damage 1: Serious eye damage/eye irritation - Category 1

· Sources

Data arise from safety data sheets, reference works and literature. ECHA: European CHemicals Agency http://echa.europa.eu (Contd. of page 8)

US ·